

SEROTONIN REUPTAKE INHIBITORS

BACKGROUND

[0001] This application claims priority to U.S. Provisional Patent Application, Serial Number 60/557,069, entitled "NOVEL SEROTONIN REUPTAKE INHIBITORS" filed on March 26, 2004, having Maria G. Miranda, Kevin G. Pinney, and James M. Dorsey, listed as the inventors, the entire content of which is hereby incorporated by reference.

[0002] This invention pertains to novel serotonin reuptake inhibitors which comprise a structural homologue of a serotonin reuptake inhibitor, known to treat depression, and a piperazine moiety, known to treat sexual dysfunction induced by serotonin-selective reuptake inhibitors (SSRIs).

[0003] Depression is a common, life-disrupting, potentially lethal illness that can affect both sexes and all ages. Untreated major depression remains a serious public health problem and its incidences are staggering. Its peak onset is in the early adult years. Suicide occurs in as many as 15% of patients with depression, especially those with recurrent episodes and hospitalizations. Therefore it becomes evident that treatment of depression is a matter of prime importance. Depression has no single cause; often, it results from a combination of factors. Whatever its cause, depression is not just a state of mind. It is related to physical changes in the brain, and connected to an imbalance of a type of chemical that carries signals in the brain and nerves. These chemicals are called neurotransmitters. Among the most important neurotransmitters related with depression are serotonin (5-HT), norepinephrine (NE), and dopamine (DA). Serotonin plays a very important role in the mood disorders, especially in anxiety and depression, aggression and impulsivity. Regulation of the mood disorders is possible either by agonistic or antagonistic action on a certain type of the serotonin receptors.

[0004] Serotonin-selective reuptake inhibitors (SSRIs), such as fluoxetine (PROZAC®), have traditionally been the mainstay of treatment for clinical depression—replacing the more toxic tricyclic antidepressants (TCAs). SSRIs have a more favorable adverse reaction profile in comparison to the TCAs and are much easier to tolerate. SSRIs exert their therapeutic effect by blocking the reuptake of serotonin into the presynaptic nerve terminal, thus increasing the synaptic concentration of serotonin. It is also believed that SSRIs increase the efficacy of the serotonin (5-HT) neurons by desensitizing 5-HT autoreceptors located on the presynaptic 5-HT nerve terminals. The ability of the 5-HT autoreceptors to inhibit the release of

5-HT decreases after long-term treatment with SSRIs, with the net effect being that a greater amount of 5-HT is released per impulse. The release of 5-HT from these neurons, thus, becomes more efficient.

[0005] Unfortunately, SSRIs are not innocuous drugs. In spite of the progress in the treatment of depression with the advent of the SSRIs, some troublesome side effects still remain—most notably, sexual dysfunction. As prescribing of SSRIs increased, the side effect of sexual dysfunction emerged, and clinicians searched for methods to treat or reverse SSRI-induced sexual dysfunction. A number of pharmacological agents—notably, drugs in the piperazine class—have been used to reverse SSRI-induced sexual dysfunction.

[0006] It is widely believed that a number of monoamine neurotransmitters have a role in the etiology of depression and, consequently, its treatment. The most notable of these neurotransmitters include norepinephrine (NE) and serotonin (5-HT). It is well known that inhibiting the reuptake of NE or 5-HT (or both) has elicited a favorable response in patients being treated for depression. Without question, however, inhibition of 5-HT reuptake has been the most widely studied, and treatment of depression has focused on inhibiting the reuptake mechanism of the neurotransmitter. Indeed, it was this premise which led to the hypothesis that altering 5-HT function could lead to an effective mechanism in the treatment of depression.

[0007] Research in the treatment of depression remains focused on 5-HT because a good number of selective 5-HT reuptake blockers tested in clinical trials have been effective in treating major depression. The general mechanism of action is hypothesized as enhanced 5-HT neurotransmission due to the increased availability of 5-HT in the synapse of these neurons (Blier et al., 1990). SSRIs generally have a 50- to 100-fold or greater selectivity for the inhibition of serotonin uptake in vitro when compared to their ability to inhibit NE or DA uptake (Lemberger et al., 1985).

[0008] Sexual dysfunction is a common problem in affective disorders such as major depression. It is important to note, however, that the drugs used in the treatment of depression can cause sexual dysfunction as well. Antidepressants such as the TCAs, monoamine oxidase inhibitors, and the SSRIs have all been reported to cause sexual side effects. The inhibitory effects of the SSRIs on sexual function are well documented, and the frequency of occurrence varies widely. By switching to a less serotonergic drug, it is surmised that the non-selective effects of 5-HT can be reduced within the synapse of the 5-HT neuron. Although the connection

between serotonergic activity and sexual function is not straightforward, it is believed that the increased central serotonergic activity of SSRIs (at the receptor subtypes 5-HT_{1A}, 5-HT_{1B}, 5-HT_{2C}, 5-HT_{2A} and 5-HT₃) plays a role in the emergence of sexual dysfunction (Modell et al., 1997). It follows, therefore, that reducing the activity of SSRIs at these receptors could allow the suppression or inhibition of the sexual response to be corrected. Strategies which have been tried in treating or reversing sexual dysfunction include: titration of an antidepressant dose, scheduling of a drug holiday during treatment, and administering a second drug (along with an SSRI, for example) to counteract the SSRI's sexual side effects.

[0009] The drugs which have been administered along with an SSRI include a broad range of medicinal agents. Drugs such as: amantadine (normally used in the treatment of Parkinsonism), cyproheptadine, buspirone, trazodone, nefazodone, methylphenidate, mianserin, yohimbine, and even sildenafil citrate have been used. Bupropion has also been of benefit. The success of each of these agents varies widely, but the best studied agents include bupropion and buspirone (Hyttel, 1982). Buspirone, which has a pyrimidinyl-piperazine group, is generally used in the treatment of generalized anxiety disorder and has been used with some success in treating patients with SSRI-induced sexual dysfunction (Gonzalez-Heydrich, 1990). It is interesting to note that other piperazine-containing groups (specifically, functionalized phenylpiperazines) have likewise been used with success in treating sexual dysfunction. These agents include: sildenafil citrate, trazodone, and nefazodone.

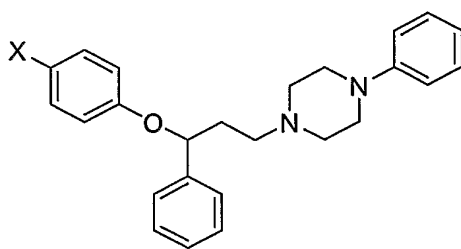
[0010] What is needed, therefore, is an SSRI with an adverse reaction profile that is devoid of sexual side effects.

SUMMARY

[0011] The present invention relates to novel serotonin reuptake inhibitors. The reuptake inhibitors comprise bi-functional organic molecules which combine serotonin transporter reuptake inhibition with serotonin (5-HT, such as 5HT_{2A}) receptor antagonism in one molecular entity. In particular, the serotonin reuptake inhibitors comprise a structural homologue of an antidepressant serotonin-selective reuptake inhibitor (SSRI) coupled to the piperidine or piperazine moiety of a compound having 5-HT receptor antagonism. The serotonin reuptake inhibitors may have a free amine group, or they may exist as salts, such as hydrochloride salts.

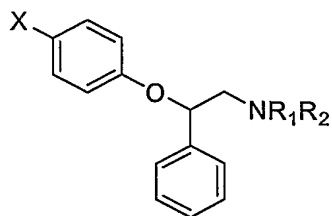
[0012] The SSRI portion of the molecule is most closely similar in structure to that of fluoxetine (Prozac®), shown in Figure 1. There are two preferred types of homologues of the SSRI portion, an aminoethane derivative (Type I) and an aminopropane derivative (Type II), as shown in Figure 2. The 5-HT receptor antagonist portion of the molecule is most closely similar in structure to the piperazinyl-containing portion of any of the compounds used to treat the side effects or sexual dysfunction caused by SSRIs. Examples are shown in Figure 3. Thus, the 5-HT receptor antagonist portion of the molecule comprises a piperazine or piperidine moiety coupled to a variety of groups, most of which are aryl groups. Figure 4 shows a generalized representation of the bi-functional molecules, in which Part a is the SSRI homologue having the ability to inhibit the serotonin reuptake transporter (SERT) and Part b is the piperazine or piperidine moiety which demonstrates antagonism to the 5-HT receptors.

[0013] One aspect of the present invention is a serotonin reuptake inhibitor having the general formula:



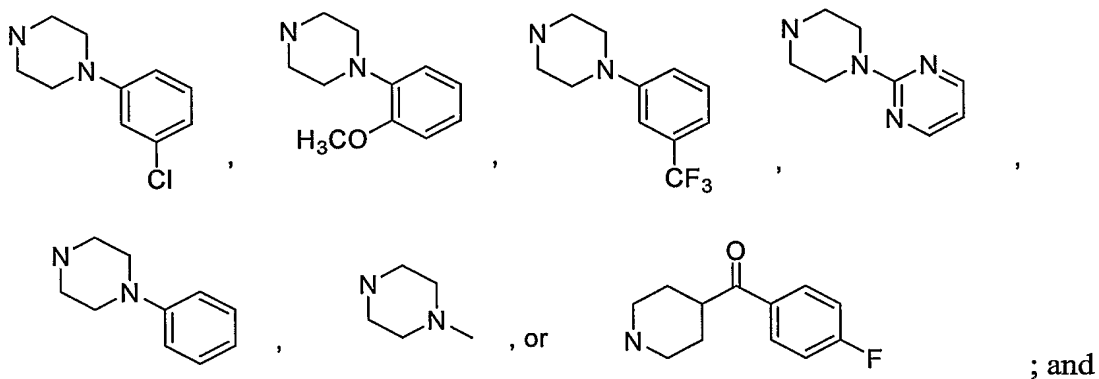
wherein X is F or CF₃; and salts thereof. This type of serotonin reuptake inhibitor is based on the Type II aminopropane derivative of the SSRI homologue.

[0014] Another aspect of the present invention is a serotonin reuptake inhibitor having the general formula:



wherein X is F or CF₃;

wherein NR₁R₂ is:



salts thereof. These types of serotonin reuptake inhibitors are based on the Type I aminoethane derivatives of the SSRI homologue.

BRIEF DESCRIPTION OF THE DRAWINGS

[0015] Figure 1 shows the structure of fluoxetine (Prozac®).

[0016] Figure 2 shows two types of SSRI homologues, an aminoethane derivative (Type I) and an aminopropane derivative (Type II).

[0017] Figure 3 shows examples of drugs traditionally administered with an SSRI to reverse or treat sexual dysfunction.

[0018] Figure 4 shows a generalized representation of the bi-functional serotonin reuptake inhibitors.

[0019] Figure 5 shows Scheme 1, a general synthetic scheme for preparing Compounds 1 – 2.

[0020] Figure 6 shows Scheme 2, a general synthetic scheme for preparing Compounds 3 – 9.

[0021] Figure 7 shows Scheme 3, a general synthetic scheme for preparing Compounds 10 – 12.

[0022] Figure 8 shows Scheme 4, a general synthetic scheme for preparing Compounds 13 – 16.

[0023] Figure 9 shows the inhibition of [³H]5-HT uptake into human platelets by Compound 10b.

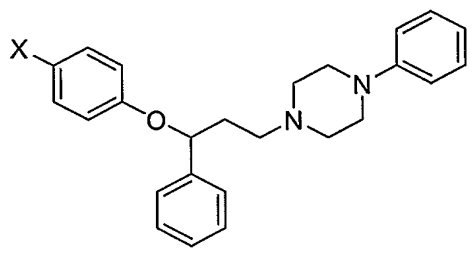
[0024] Figure 10 shows the inhibition of [³H]5-HT uptake into human platelets by Compound 11b.

[0025] Figure 11 shows the inhibition of [³H]5-HT uptake into human platelets by Compound 12b.

DETAILED DESCRIPTION OF PREFERRED EMBODIMENTS

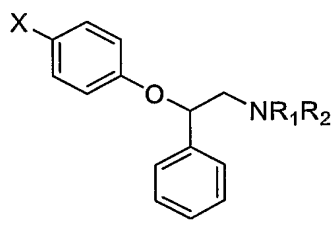
[0026] The present invention relates to serotonin reuptake inhibitors comprising an SSRI homologue coupled to a piperazine or piperidine moiety. The serotonin reuptake inhibitors are bi-functional organic molecules which combine 5-HT_{2A} receptor antagonism and serotonin reuptake inhibition into one molecular entity. The SSRI homologue shows an affinity to the serotonin reuptake transporter (SERT) and has antidepressant properties. The piperazine or piperidine moiety demonstrates an affinity to 5-HT receptors and restores the undesired side effects of SSRIs.

[0027] One aspect of the present invention is a serotonin reuptake inhibitor having the general formula:



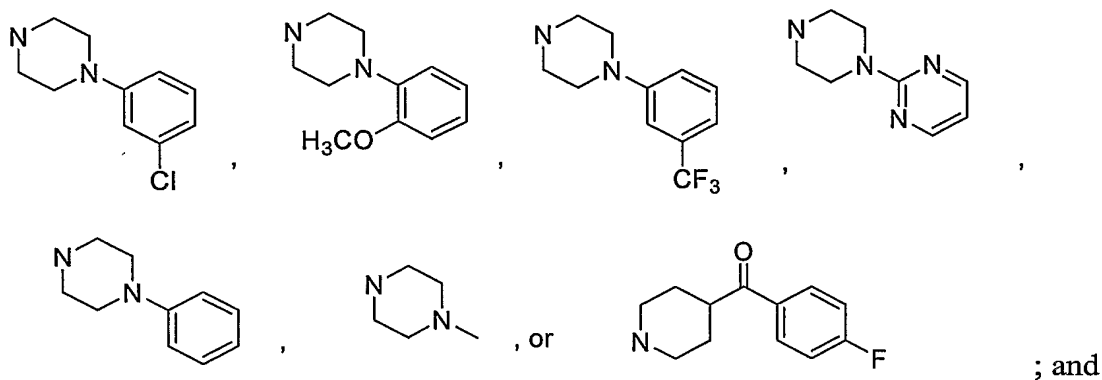
wherein X is F or CF₃; and salts thereof.

[0028] Another aspect of the present invention is a serotonin reuptake inhibitor having the general formula:



wherein X is F or CF₃;

wherein NR₁R₂ is:



salts thereof.

[0029] Preferred examples of the serotonin reuptake inhibitors were prepared according to the general synthetic schemes shown in Schemes 1 – 4 in Figures 5 – 8.

[0030] The serotonin reuptake inhibitors, and their salts, exhibit single-site binding at the site of the serotonin reuptake transporter (SERT) and inhibition of 5-HT receptor uptake.

EXAMPLE 1. GENERAL METHODS

[0031] Chemicals used in synthesis were obtained from the Aldrich Chemical Company (St. Louis, MO), Fisher Scientific (Hampton, NH) and ACROS Chemicals (Belgium). Tetrahydrofuran (THF), dried with potassium and benzophenone, and methylene chloride, dried with calcium hydride, were freshly distilled immediately prior to use. Dimethylformamide (DMF) was purchased as an anhydrous commercial product in a SureSeal® container from the Aldrich Chemical Company. Silica gel plates for analytical thin layer chromatography (TLC) were obtained from EM Science (Gibbstown, NJ). Column chromatography was performed using silica gel (300-400 mesh) obtained from Merck EM Science. ^1H and ^{13}C NMR were recorded in deuterated chloroform (CDCl_3) using an AMX 360 MHz (90 MHz for ^{13}C) and/or DPX Avance 300 MHz (75 MHz for ^{13}C) Bruker NMR spectrometers. Peaks are listed as broad (b), singlet (s), doublet (d), triplet (t), quartet (q), or multiplet (m), with the coupling constant (J) expressed in hertz (Hz). High resolution mass spectra were recorded using a VG/Fisons High Resolution GC/Mass Spectrometer using perfluorokerosene (PFK) as the calibration standard. All elemental analyses were conducted by Atlantic Microlab, Inc. in Norcross, Georgia. Melting points were determined in open capillary tubes using a Thomas-Hoover electronically heated melting point apparatus and are uncorrected.

EXAMPLE 2. SYNTHESIS OF COMPOUNDS 1 AND 2

[0032] The following compounds were synthesized according to Scheme 1 shown in Figure 5 and the procedures described below:

Cpd	Structure (Free Amine)	Cpd	Structure (Hydrochloride Salt)
1a		1b	
2a		2b	

1. Preparation of 1-Phenyl-3-(4-phenyl-piperazin-1-yl)-propan-1-one (19)

[0033] A well-stirred mixture of acetophenone (**17**) (3.40 mL, 29.0 mmol), 1-phenylpiperazine hydrochloride (**18**) (4.48g, 22.5 mmol), and 0.05mL of concentrated HCl in absolute ethanol (35mL) was heated under N₂ at reflux conditions. Paraformaldehyde (1.00g, 33.3 mmol) was added in four equal portions over 40 minutes. The reaction mixture was further refluxed for 6 hours. The solution was then cooled and poured onto crushed ice. The solid was separated by filtration, dried, and re-crystallized using a small amount of aqueous ethanol (<10mL). The solution was allowed to cool overnight and yielded 5.32 g the desired product **19** (18.1 mmol, 62%) as a white, yellow crystal. ¹H-NMR (CDCl₃): δ 8.03 (2H, d, J=7.24 Hz), 7.63 (1H, t, J=7.37, 7.42 Hz), 7.50 (2H, t, J=7.87, 7.35 Hz), 7.31 (2H, m), 6.97 (3H, m), 3.89 (2H, t, J=6.89, 6.72 Hz), 3.62 (8H, m), 3.10 (2H, m). ¹³C-NMR (CDCl₃): δ 195.99, 149.27, 135.41, 134.24, 129.51, 128.37, 122.00, 117.43, 52.36, 51.97, 46.91, 33.14. Melting Point: 182-185°C.

2. Preparation of 1-Phenyl-3-(4-phenyl-piperazin-1-yl)-propan-1-ol (20)

[0034] To a well-stirred solution of 1-Phenyl-3-(4-phenyl-piperazin-1-yl)-propan-1-one (**19**) (5.32g, 18.1 mmol) in 25mL of methanol under N₂, sodium borohydride, NaBH₄,

(1.722g, 45.5 mmol) was added over a period of 15 minutes at a temperature between 10 to 15°C. Care was taken to limit exposure of the NaBH₄ to the air. The reaction was placed at room temperature after the solution stopped bubbling and was allowed to continue stirring for 4-7 hours. The reaction mixture was poured onto crushed ice and treated with glacial acetic acid (0.2mL). It was then extracted with ethyl acetate, EtOAc, (20mL x 4), and washed with water (10mL x 3). It was dried with anhydrous sodium sulfate and recollected by filtration. The solvent was removed under reduced pressure, and the residue obtained was triturated with methanol (3mL). The solution remained in the refrigerator overnight. The crystallized solid was filtered and dried given 3.57 g the desired product **20** (12.0 mmol, 66.7%) as a white crystalline solid. ¹H-NMR (CDCl₃): δ 7.37 (4H, m), 7.27 (3H, m), 6.92 (3H, m), 4.97 (1H, t, J=5.68, 5.75 Hz), 3.25 (4H, m), 2.72 (7H, m), 1.92 (2H, m). ¹³C-NMR (CDCl₃): δ 151.09, 144.78, 129.18, 128.07, 127.00, 125.55, 120.06, 116.29, 75.53, 57.10, 53.28, 49.29, 33.72. Melting Point: 85-87°C.

3. Preparation of 1-[3-(4-Fluoro-phenoxy)-3-phenyl-propyl]-4-phenyl-piperazine (1a)

[0035] In a stirring solution of THF (10 mL) under N₂, 2 equivalents of 4-fluorophenol (1.40g, 12.4 mmol) was added to a solution of 1-Phenyl-3-(4-phenyl-piperazin-1-yl)-propan-1-ol (**20**) (1.84g, 6.22 mmol) and allowed to dissolve. 1.5 equivalents of triphenylphosphine, PPh₃, (2.49g, 9.32 mmol) was then added and dissolved. After the reaction was allowed to equilibrate at a temperature between 10-15°C, 1.5 equivalents of diisopropyl azodicarboxylate, DIAD, (1.85 mL, 9.32 mmol) was added dropwise. The reaction was then allowed to return to room temperature and continued stirring for 48 hours. The solvent was then evaporated under reduced pressure and aspirated under vacuum for an extended period. Purification was performed by Flash Column Chromatography on silica gel by first washing with 500mL of 10% EtOAc/hexanes and then extracting with 20% EtOAc/hexanes. Analytical thin-layer chromatography showed that the desired product had an R_f value of approximately 0.56 when developed in 30% EtOAc/hexanes. The fractions containing the desired product were combined, and the solvent was removed under reduced pressure. 0.70 g of the product **1a** (1.795mmol, 29%) was obtained as a yellow, clear oil. ¹H-NMR (CDCl₃): δ 7.29 (7H, m), 6.93 (2H, d, J= 7.93 Hz), 6.86 (5H, m), 5.16 (1H, dd, J=5.00, 4.96 Hz), 3.21 (4H, t, J=4.85, 5.10 Hz), 2.60 (6H, m), 2.23 (1H, sextuplet, J=7.26, 6.75, 7.44, 6.91, 7.72 Hz), 2.02 (1H, m). ¹³C-NMR (CDCl₃): δ 155.62, 154.34, 151.28, 141.61, 129.11, 128.63, 127.69, 126.02, 119.72, 117.15, 117.04, 116.09, 115.80, 115.50, 79.34, 54.62, 53.26, 49.15, 35.96. ¹⁹F-NMR (CDCl₃): δ -123.77.

Elemental Analysis (%) calculated for $C_{25}H_{27}ON_2F$: C, 76.92; H, 6.92; O, 4.10; N, 7.18; F, 4.87. Found: C, 77.17; H, 7.28; O, N/A; N, 6.85; F, N/A.

4. Preparation of 1-Phenyl-4-[3-phenyl-3-(4-trifluoromethyl-phenoxy)-propyl]-piperazine (2a)

[0036] In a stirring solution of THF (10 mL) under N_2 , 2 equivalents of 4-trifluoromethyl phenol (1.873g, 11.6 mmol) was added to a solution of 1-Phenyl-3-(4-phenyl-piperazin-1-yl)-propan-1-ol (**20**) (1.71g, 5.78 mmol) and allowed to dissolve. 1.5 equivalents of triphenylphosphine, PPh_3 , (2.293 g, 8.67 mmol) was then added and dissolved. After the reaction was allowed to equilibrate at a temperature between 10-15°C, 1.5 equivalents of diisopropyl azodicarboxylate, DIAD, (1.7mL, 8.67 mmol) was added dropwise. The reaction was then allowed to return to room temperature and continued stirring for 48 hours. The solvent was then evaporated under reduced pressure and aspirated for an extended period. Purification was performed by Flash Column Chromatography on silica gel by first washing with 500mL of 10% EtOAc/hexanes and then extracting with 20% EtOAc/hexanes. Analytical thin-layer chromatography showed that the desired product had an R_f value of approximately 0.56 when developed in 30% EtOAc/hexanes. The fractions containing the desired product were combined, and the solvent was removed under reduced pressure. 1.38 g of the product **2a** (3.17 mmol, 54%) was obtained as a yellow-orange oil. 1H -NMR ($CDCl_3$): δ 7.43 (2H, d, $J=8.56$ Hz), 7.35 (4H, m), 7.27 (3H, m), 6.91 (5H, m), 5.28 (1H, dd, $J=4.90, 4.90$ Hz), 3.23 (4H, t, $J=4.89, 5.09$ Hz), 2.66 (4H, t, $J=5.87, 4.00$ Hz), 2.60 (2H, t, $J=7.59, 7.13$ Hz), 2.27 (1H, sextuplet, $J=7.34, 7.73, 6.48, 7.11, 7.58$ Hz), 2.07 (1H, m). ^{13}C -NMR ($CDCl_3$): δ 160.56, 151.12, 140.83, 129.15, 128.81, 127.94, 127.14, 126.75, 125.85, 119.97, 116.16, 115.76, 115.56, 78.62, 54.50, 53.21, 49.03, 35.72. ^{19}F -NMR ($CDCl_3$): δ -61.52. Elemental Analysis (%) calculated for $C_{26}H_{27}ON_2F_3$: C, 70.91; H, 6.14; O, 3.64; N, 6.36; F, 12.95. Found: C, 70.94; H, 6.33; O, N/A; N, 6.17; F, N/A.

5. Preparation of the Hydrochloride salt 1b from amine 1a

[0037] To a solution of 0.164g of product **1a** (0.4mmol) in 10 mL of stirring THF under N_2 , 10 equivalents of 2M HCl in anhydrous diethyl ether (2mL, 4 mmol) was added at room temperature. The mixture was allowed to stir for 12 to 16 hours. The reaction mixture was then filtered and redissolved in 5mL of anhydrous ethyl alcohol. The reaction was heated to reflux conditions for 5 to 10 minutes, and the solvent was then evaporated by reduced pressure yielding 0.083 g of the final product **1b** (0.19mmol, 46%) as a blue-green crystal. The

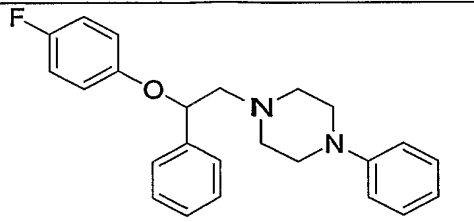
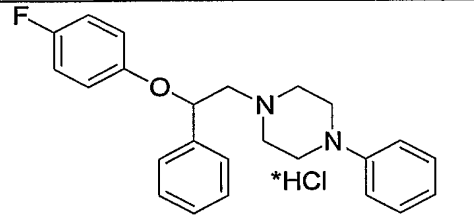
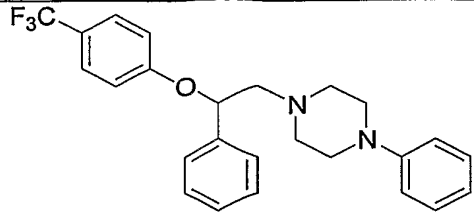
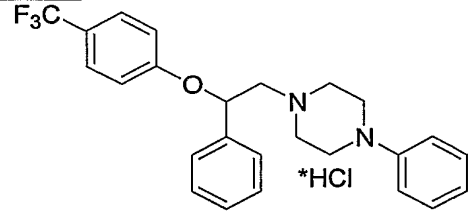
hydrochloride salt product was confirmed by treating a sample of the product with strong base, extracting with diethyl ether, evaporating the organic layer under reduced pressure, and confirming by NMR in CDCl_3 . The NMR data corresponded with the free amine product **1a**. Melting Point: 176-179°C.

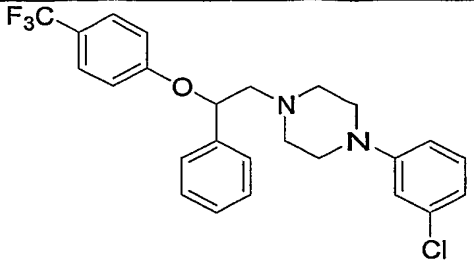
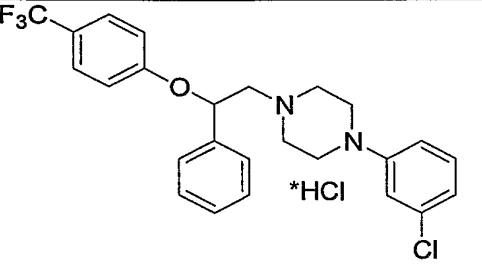
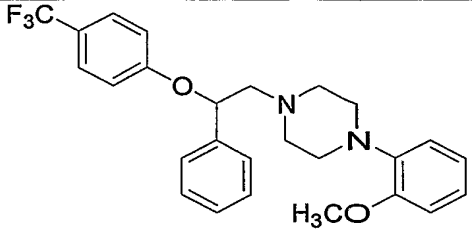
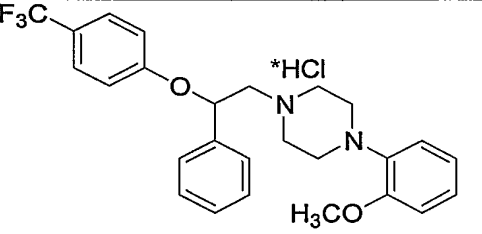
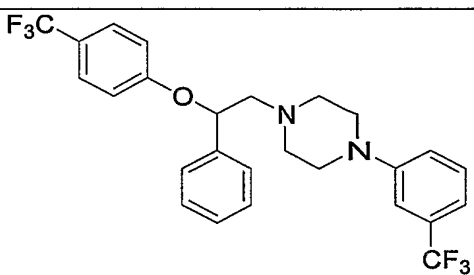
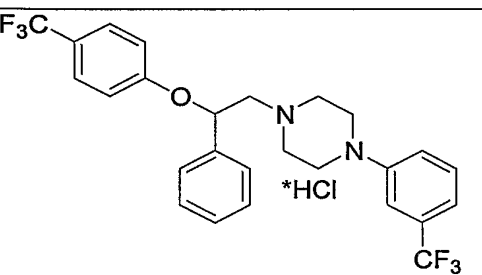
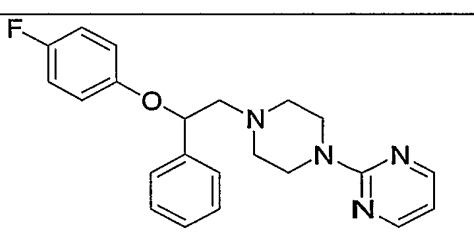
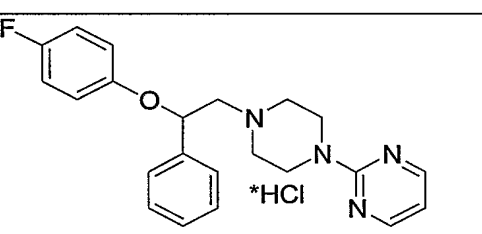
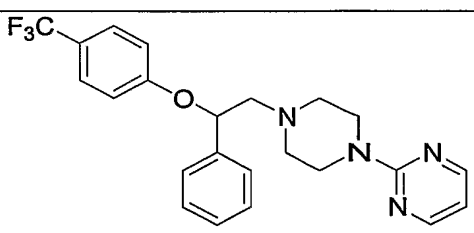
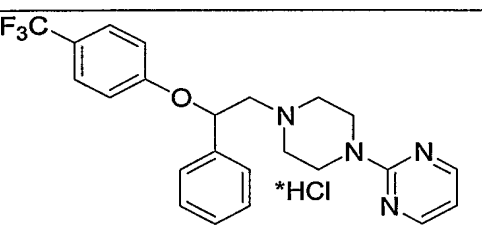
6. Preparation of the Hydrochloride salt 2b from amine 2a

[0038] To a solution of 0.17 g of product **2a** (0.4 mmol) in 10 mL of stirring THF under N_2 , 10 equivalents of 2M HCl in anhydrous diethyl ether (2 mL, 4 mmol) was added at room temperature. The mixture was allowed to stir for 12 to 16 hours. The reaction mixture was then filtered and redissolved in 5 mL of anhydrous ethyl alcohol. The reaction was heated to reflux conditions for 5 to 10 minutes, and the solvent was then evaporated by reduced pressure yielding 0.077g of the final product **2b** (0.26 mmol, 42%) as a white-yellow crystal. The hydrochloride salt product was confirmed by treating a sample of the product with strong base, extracting with diethyl ether, evaporating the organic layer under reduced pressure, and confirming by NMR in CDCl_3 . The NMR data corresponded with the free amine product **2a**. Melting Point: 180-183°C

EXAMPLE 3. SYNTHESIS OF COMPOUNDS 3 – 9

[0039] The following compounds were synthesized according to Scheme 2 shown in Figure 6 and the procedures described below:

Cpd	Structure (Free Amine)	Cpd	Structure (Hydrochloride Salt)
3a		3b	
4a		4b	

5a		5b	
6a		6b	
7a		7b	
8a		8b	
9a		9b	

1. General procedure for preparation of free amines: Preparation of 1-Phenylpiperazine (I)

[0040] 1.05 g of 1-Phenylpiperazine hydrochloride (5.20 mmol) was added to a well stirred solution of de-ionized water (20mL) and excess solution of 2M NaOH (100mL). After stirring for 10 minutes under N₂ gas, the product was extracted with diethyl ether (30 mL x 4). It was dried with anhydrous sodium sulfate, filtered, and the solvent was evaporated yielding 0.84 g

of the desired free amine 1-phenylpiperazine **I** (5.19 mmol, 98%). $^1\text{H-NMR}$ (CDCl_3): δ 7.27 (2H, m), 6.93 (2H, dd, $J=1.07$, 0.98 Hz), 6.86 (1H, tt, $J=1.06$, 1.04, 0.89, 0.89, 1.09, 1.06 Hz), 3.13 (4H, m), 3.04 (4H, m), 1.88 (1H, bs).

2. General Procedure for Preparation of 1-Phenyl-2-(aryl-piperazin-1-yl)-ethanones (22 to 26): Preparation of 1-Phenyl-2-(4-phenyl-piperazin-1-yl)-ethanone (22)

[0041] 6.10 g of potassium carbonate, K_2CO_3 , (6.10 mmol) was added to a dissolved, stirring solution of 1-phenylpiperazine (**I**) (0.84 g, 5.19 mmol) in 20 mL of acetonitrile under N_2 . After 5 minutes, 0.80 g of 2-chloroacetophenone (**21**) (5.19 mmol) was added. After 5 more minutes, the solution was brought to reflux conditions (approximately 90°F) and left for 16 hours. The solution was then allowed to cool and 50 mL of water was added. The organic layer was then extracted with dichloromethane (50 mL x 3) and washed with brine solution (50mL). It was dried with anhydrous sodium sulfate, filtered, and the solvent was evaporated. A Flash Chromatography Column was performed with 30% EtOAc/hexanes to separate the desired product. The fractions containing the product were combined, and the solvent was removed under reduced pressure. 0.87 g of the product **22** (3.107 mmol, 60%) was obtained as a yellow solid. $^1\text{H-NMR}$ (CDCl_3): δ 8.03 (2H, d, $J=8.32$ Hz), 7.59 (1H, t, $J=7.18$, 7.57 Hz), 7.47 (2H, m), 7.28 (2H, m), 6.95 (2H, d, $J=7.88$ Hz), 6.87 (1H, t, $J=7.29$, 7.26 Hz), 3.91 (2H, s), 3.29 (4H, t, $J=4.93$, 5.05 Hz), 2.80 (4H, t, $J=5.13$, 4.82 Hz). $^{13}\text{C-NMR}$ (CDCl_3): δ 196.03, 151.21, 135.98, 133.38, 129.13, 128.62, 128.12, 117.19, 116.19, 64.26, 53.49, 49.00. Melting Point: 104-106°C.

3. General Procedure for Preparation of 1-Phenyl-2-(aryl-piperazin-1-yl)-ethanols (27 to 31): Preparation of 1-Phenyl-2-(4-phenyl-piperazin-1-yl)-ethanol 27

[0042] To a well-stirred solution of 1-Phenyl-2-(4-phenyl-piperazin-1-yl)-ethanone (**22**) (0.87g, 3.107mmol) in 25mL of methanol under N_2 , 2.5 equivalents of sodium borohydride, NaBH_4 , (0.294g, 7.768 mmol) was added over a period of 15 minutes at a temperature of 10 to 15°C. Care was taken to limit exposure of the NaBH_4 to the air. The reaction was placed at room temperature after the solution stopped bubbling and was allowed to continue stirring for 4-7 hours. The reaction mixture was poured onto crushed ice and treated with glacial acetic acid (0.2mL). It was then extracted with ethyl acetate, EtOAc, (20mL x 4), and washed with water (10mL x 3). It was dried with anhydrous sodium sulfate and recollected by filtration. The crystallized solid was aspirated given 0.84g of the desired product **27** (2.979 mmol, 96%) as a white crystalline solid. $^1\text{H-NMR}$ (CDCl_3): δ 7.39 (4H, m), 7.28 (3H, m), 6.95 (2H, dd, $J=0.99$,

0.83 Hz), 6.88 (1H, t, J=7.27, 7.30 Hz), 4.80 (1H, dd, J=4.67, 4.68 Hz), 3.25 (4H, dt, J=3.91, 2.47, 2.57, 3.67 Hz), 2.95 (2H, m), 2.61 (4H, m). ^{13}C -NMR (CDCl_3): δ 151.16, 141.92, 129.17, 128.41, 127.60, 125.86, 119.96, 116.19, 68.78, 66.20, 53.06, 49.33. Melting Point: 107-109°C.

4. General Procedure for Preparation of 1-[2-(4-aryloxy)-2-phenyl-ethyl]-4-aryl-piperazines (3a to 9a):

a. Preparation of 1-[2-(4-Fluoro-phenoxy)-2-phenyl-ethyl]-4-phenyl-piperazine (3a)

[0043] In a stirring solution of THF (20mL) under N_2 , 2 equivalents of 4-trifluoromethylphenol (0.564g, 5.035 mmol) was added to a solution of 1-Phenyl-2-(phenyl-piperazin-1-yl)-ethanol (**27**) (0.71 g, 2.518 mmol) and allowed to dissolve. 1.5 equivalents of triphenylphosphine, PPh_3 , (1.00 g, 4.574 mmol) was then added and dissolved. After the reaction was allowed to equilibrate at a temperature between 10-15°C, 1.5 equivalents of diisopropyl azodicarboxylate, DIAD, (0.75 mL, 4.574 mmol) was added dropwise. The reaction was then allowed to return to room temperature and continued stirring for 48 hours. The solvent was then evaporated under reduced pressure and aspirated for an extended period. Purification was performed by Flash Column Chromatography on silica gel by first washing with 500mL of 10% EtOAc/hexanes and then extracting with 20% EtOAc/hexanes. Analytical thin-layer chromatography showed that the desired product had an R_f value of approximately 0.50 when developed in 30% EtOAc/hexanes. The fractions containing the desired product were combined, and the solvent was removed under reduced pressure. 0.80 g of the product **3a** (2.128 mmol, 36%) was obtained as a white solid. ^1H -NMR (CDCl_3): δ 7.35 (4H, m), 7.26 (3H, m), 6.95 (2H, dd, J=0.92, 0.81 Hz), 6.84 (5H, m), 5.29 (1H, dd, J=3.10, 3.09 Hz), 3.21 (4H, t, J=5.04, 4.99 Hz), 3.06 (1H, dd, J=8.48, 8.49 Hz), 2.85 (2H, m), 2.74 (3H, m). ^{13}C -NMR (CDCl_3): δ 155.72, 153.92, 151.31, 140.35, 129.10, 128.67, 127.68, 126.13, 119.71, 117.28, 117.17, 116.05, 115.85, 115.54, 79.71, 65.39, 53.63, 49.24. ^{19}F -NMR (CDCl_3): δ -123.53. Melting Point: 80-82°C. Elemental Analysis (%) calculated for $\text{C}_{24}\text{H}_{25}\text{ON}_2\text{F}$: C, 76.60; H, 6.65; O, 4.25; N, 7.45; F, 5.05. Found: C, 76.58; H, 6.75; O, N/A; N, 7.46; F, N/A.

b. Preparation of 1-Phenyl-4-[2-phenyl-2-(4-trifluoromethyl-phenoxy)-ethyl]-piperazine (4a)

[0044] In a stirring solution of THF (20 mL) under N_2 , 2 equivalents of 4-fluorophenol (0.965g, 5.985 mmol) was added to a solution of 1-Phenyl-2-(phenyl-piperazin-1-yl)-ethanol (**27**) (0.84 g, 2.979 mmol) and allowed to dissolve. 1.5 equivalents of

triphenylphosphine, PPh_3 , (1.172 g, 4.468 mmol) was then added and dissolved. After the reaction was allowed to equilibrate at a temperature between 10-15°C, 1.5 equivalents of diisopropyl azodicarboxylate, DIAD, (0.88 mL, 4.468 mmol) was added dropwise. The reaction was then allowed to return to room temperature and continued stirring for 48 hours. The solvent was then evaporated under reduced pressure and aspirated for an extended period. Purification was performed by Flash Column Chromatography on silica gel by first washing with 500 mL of 10% EtOAc/hexanes and then extracting with 20% EtOAc/hexanes. Analytical thin-layer chromatography showed that the desired product had an R_f value of approximately 0.48 when developed in 30% EtOAc/hexanes. The fractions containing the desired product were combined, and the solvent was removed under reduced pressure. 0.51g of the product **4a** (1.197 mmol, 40%) was obtained as a clear yellow oil. ^1H -NMR (CDCl_3): δ 7.44 (2H, d, $J=8.45$ Hz), 7.36 (4H, m), 7.27 (3H, m), 6.92 (4H, m), 6.85 (1H, t, $J=7.28, 7.30$ Hz), 5.42 (1H, dd, $J=3.19, 3.20$ Hz), 3.19 (4H, t, $J=5.04, 4.99$ Hz), 3.08 (1H, dd, $J=8.33, 8.36$ Hz), 2.78 (5H, m). ^{13}C -NMR (CDCl_3): δ 160.25, 151.30, 139.74, 129.12, 128.82, 128.06, 126.82, 126.77, 126.72, 126.01, 123.23, 119.78, 116.09, 115.93, 79.22, 65.66, 53.30, 49.27. ^{19}F -NMR (CDCl_3): δ -61.53. Elemental Analysis (%) calculated for $\text{C}_{25}\text{H}_{25}\text{ON}_2\text{F}_3$: C, 70.42; H, 5.87; O, 3.76; N, 6.57; F, 13.38. Found: C, 70.51; H, 5.94; O, N/A; N, 6.57; F, N/A.

5. General Procedure for Preparation of Hydrochloride Salts (3b to 9b) from 3a to 9a

a. Preparation of the Hydrochloride salt 3b from the free amine product 3a

[0045] To a solution of 0.80 g of product **3a** (2.12mmol) in 10 mL of stirring THF under N_2 , 10 equivalents of 2M HCl in anhydrous diethyl ether (11mL, 21.2 mmol) was added at room temperature. The mixture was allowed to stir for 12 to 16 hours. The reaction mixture was then filtered and redissolved in 10 mL of anhydrous ethyl alcohol. The reaction was heated to reflux conditions for 5 to 10 minutes, and the solvent was then evaporated by reduced pressure yielding 0.66 g of the final product **3b** (1.602mmol, 75%) as a white-cream solid. The hydrochloride salt product was confirmed by treating a sample of the product with strong base, extracting with diethyl ether, evaporating the organic layer under reduced pressure, and confirming by NMR in CDCl_3 . The NMR data corresponded with the free amine product **3a**. Melting Point: 195-198°C.

b. Preparation of the Hydrochloride salt 4b from the free amine product 4a

[0046] To a solution of 0.96 g of product **4a** (2.25mmol) in 15mL of stirring THF under N₂, 10 equivalents of 2M HCl in anhydrous diethyl ether (12mL, 22.5 mmol) was added at room temperature. The mixture was allowed to stir for 12 to 16 hours. The reaction mixture was then filtered and redissolved in 10mL of anhydrous ethyl alcohol. The reaction was heated to reflux conditions for 5 to 10 minutes, and the solvent was then evaporated by reduced pressure yielding 0.85 g of the final product **4b** (1.839mmol, 82%) as a blue-green solid. The hydrochloride salt product was confirmed by treating a sample of the product with strong base, extracting with diethyl ether, evaporating the organic layer under reduced pressure, and confirming by NMR in CDCl₃. The NMR data corresponded with the free amine product **4a**. Melting Point: 161-163°C.

EXAMPLE 4. SYNTHESIS OF COMPOUNDS 10 – 12

[0047] The following compounds were synthesized according to Scheme 3 shown in Figure 7 and the procedures described below:

Cpd	Structure (Free Amine)	Cpd	Structure (Hydrochloride Salt)
10a		10b	
11a		11b	
12a		12b	

1. Preparation of (2-hydroxy-2-phenylethyl)-carbamic acid *tert*-butyl ester (33)

[0048] To a well-stirred solution of 2-amino-1-phenylethanol (**32**) (5.05 g, 36.8 mmol) in anhydrous DMF (50 mL) under N₂ at 25 °C was added 1.5 molar equivalents of di-*tert*-butyl dicarbonate (*t*-boc anhydride) (13.24 g, 60.7 mmol). After dissolution of the *t*-boc anhydride, the mixture was brought to 50 °C for approximately 15 minutes and then allowed to cool to 25 °C. After stirring overnight (8-12 hours), ethyl acetate (EtOAc) was added to the reaction mixture. Water was then added, and the organic layer was extracted three times with water and subsequently dried over anhydrous magnesium sulfate (MgSO₄). The mixture was then filtered and the solvent removed under reduced pressure. The resulting residue was suspended in a minimal amount of EtOAc and heated to 55-60 °C until completely dissolved. The mixture was allowed to cool overnight, remaining undisturbed, and afforded the desired product **33** (6.73 g, 28.4 mmol, 78%) as translucent, slightly yellow crystals. ¹H-NMR (CDCl₃): δ 7.31 (5H, m), 4.97 (1H, bs), 4.82 (1H, dd, *J* = 3.9, 3.7 Hz), 3.47 (1H, ddd, *J* = 14.0, 6.5, 3.2 Hz), 3.27 (1H, dd, *J* = 7.8, 5.4 Hz), 3.22 (1H, dd, *J* = 8.6, 6.1 Hz), 1.71 (1H, bs), 1.44 (9H, s). ¹³C-NMR (CDCl₃): δ 157.0 (faint), 142.2, 128.9, 128.9, 128.2, 126.3, 126.3, 80.3, 74.4, 48.8, 28.8, 28.8, 28.8. Melting point: 123-124 °C.

2. Preparation of [2-(4-fluorophenoxy)-2-phenylethyl]-carbamic acid *tert*-butyl ester (34)

[0049] To a well-stirred solution of **33** (6.73 g, 28.4 mmol) in anhydrous THF (40 mL) under N₂ at 25 °C is added 2 molar equivalents (eq.) of 4-fluorophenol (6.55 g, 58.4 mmol). After complete dissolution of the 4-fluorophenol, 1.5 eq. of triphenylphosphine (PPh₃) (11.68 g, 44.5 mmol) was added to the mixture and allowed to completely dissolve. The reaction vessel was then cooled in an ice bath to 0 °C. After 20 minutes, 1.5 eq. of diethylazodicarboxylate (DEAD) (7.96 g, 45.7 mmol) was slowly added dropwise to the stirring mixture. The reaction vessel was subsequently allowed to return to room temperature (25 °C). After stirring overnight (8-12 hours), the triphenylphosphine salts formed during the reaction were removed by filtration and the solvent evaporated under reduced pressure. The residue was adsorbed onto silica gel and purified by flash column chromatography (silica gel, 5:95 EtOAc/hexanes) which afforded the desired intermediate **34** (4.04 g, 12.2 mmol, 43%) as a white solid. ¹H-NMR (CDCl₃): δ 7.34 (1H, d, *J* = 5.4 Hz), 7.33 (1H, d, *J* = 8.0 Hz), 7.31 (1H, d, *J* = 4.8 Hz), 7.30 (1H, d, *J* = 7.1 Hz), 7.27 (1H, d, *J* = 4.2 Hz), 6.89 (1H, d, *J* = 10.6 Hz), 6.85 (1H, d, *J* = 9.0 Hz), 6.78 (1H, d, *J* = 9.2

Hz), 6.76 (1H, d, $J = 9.2$ Hz), 5.16 (1H, dd, $J = 8.5, 3.4$ Hz), 5.02 (1H, bs), 3.63 (1H, ddd, $J = 14.4, 8.0, 3.7$ Hz), 3.39 (1H, ddd, $J = 13.3, 8.3, 4.6$ Hz), 1.43 (9H, s). ^{13}C -NMR (CDCl_3): δ 158.7, 156.0, 155.8, 153.8, 138.5, 128.7, 128.1, 128.1, 126.1, 126.1, 117.0, 116.9, 115.9, 115.6, 80.0, 68.1, 47.2, 28.4, 28.4, 28.4. ^{19}F -NMR (CDCl_3): δ -123.8. Melting point: 98-102 °C.

3. Preparation of 2-(4-fluorophenoxy)-2-phenyl-ethylamine (35)

[0050] A 30 mL volume of 4 N HCl in dioxane was prepared by adding dropwise 10 mL of 12 N (i.e.; concentrated) hydrochloric acid to 20 mL of well-stirred amount of 1,4-dioxane. The t-boc protected 4-fluorophenyl ether intermediate **34** (4.04 g, 12.2 mmol) was dissolved in a minimal amount of 1,4-dioxane. The 4 N HCl was then added dropwise to the reaction vessel at 25 °C over approximately a 10 minute period. After 90 minutes, a saturated solution of sodium bicarbonate (NaHCO_3) was added to the reaction mixture. The aqueous layer was washed with three separate 30 mL portions of CH_2Cl_2 . The combined organic layers were then dried over anhydrous MgSO_4 and filtered. Purification was performed by flash column chromatography (on silica gel) starting with 100% EtOAc until any remaining starting material had eluted; then the polarity of the eluent was increased using 30:70 methanol/EtOAc. Analytical thin-layer chromatography showed that the desired product had an R_f value of approximately 0.4 when developed in 30:70 methanol/EtOAc. Fractions containing the desired product were combined and the solvent was removed under reduced pressure. The resultant residue after solvent evaporation was re-suspended in CH_2Cl_2 and filtered. The solvent from the filtrate was evaporated under reduced pressure and afforded the desired free amine **5** (1.80 g, 7.8 mmol, 64%) as a slightly yellow oil. ^1H -NMR (CDCl_3): δ 7.28 (5H, m), 6.86 (1H, d, $J = 9.3$ Hz), 6.83 (1H, d, $J = 8.0$ Hz), 6.79 (1H, d, $J = 9.3$ Hz), 6.78 (1H, d, $J = 9.4$ Hz), 5.06 (1H, dd, $J = 7.1, 4.4$ Hz), 3.14 (1H, d, $J = 6.8$ Hz), 3.12 (1H, d, $J = 7.7$ Hz), 2.35 (2H, bs). ^{13}C -NMR (CDCl_3): δ 158.6, 156.0, 154.0, 139.0, 128.7, 128.0, 126.1, 126.1, 117.0, 116.9, 115.8, 115.6, 82.3, 49.0. ^{19}F -NMR (CDCl_3): δ -124.0.

4. 2-(4-fluorophenoxy)-2-phenyl-ethyl bromide (36)

[0051] To a rapidly stirred solution of 1.1 eq. of TiBr_4 (3.15 g, 8.67 mmol) in 40 mL of DMF at 25 °C under N_2 was added dropwise 1.1 eq. of t-butyl nitrite (1.2 mL of 90% t-butyl nitrite, $d = 0.86$, 8.6 mmol) dissolved in approx. 5 mL of DMF. Upon addition of the t-butyl nitrite, a color change from reddish orange to orange-yellow was observed. The primary amine starting material **35** (1.80 g, 7.8 mmol) was dissolved in approximately 5 mL of DMF and added

dropwise to the reaction flask over a 20-minute period. Gas evolution was observed during addition of the amine starting material and was complete within approximately 5 minutes following its complete addition. After complete gas evolution, the reaction mixture was added to 150 mL of 20% aqueous hydrochloric acid (20% aq. HCl) and extracted with four 30 mL portions of ethyl ether. The combined organic layers were subsequently dried over anhydrous MgSO_4 . The solvent was then removed under reduced pressure. Analytical thin-layer chromatography (TLC) indicated that the desired product had an R_f value of approx. 0.85 when developed in 30:70 EtOAc/hexanes. Purification was performed using flash column chromatography (silica gel, 100% hexanes) to isolate the desired bromide product **36** (0.74 g, 2.6 mmol, 33%) as a yellow oil. $^1\text{H-NMR}$ (CDCl_3): δ 7.37 (5H, m), 6.89 (1H, d, $J = 8.0$ Hz), 6.87 (1H, d, $J = 8.0$ Hz), 6.83 (1H, d, $J = 4.5$ Hz), 6.81 (1H, d, $J = 4.6$ Hz), 5.22 (1H, dd, $J = 8.4, 4.0$ Hz), 3.72 (1H, dd, $J = 10.9, 8.4$ Hz), 3.61 (1H, dd, $J = 10.9, 4.0$ Hz). $^{13}\text{C-NMR}$ (CDCl_3): δ 159.6, 156.5, 154.3, 154.3, 138.9, 129.3, 129.2, 126.8, 118.0, 117.9, 116.4, 116.1, 81.7, 36.4. $^{19}\text{F-NMR}$ (CDCl_3): δ -123.2.

5. Preparation of 1-(3-chlorophenyl)-4-[2-(4-fluorophenoxy)-2-phenylethyl]-piperazine (10a)

[0052] To a well-stirred solution of approx. 1.1 eq. of 1-(3-chlorophenyl)-piperazine (**II**) (0.26 g, 1.3 mmol) in 20mL DMF at 25 °C under N_2 and fitted with a reflux condenser was added a minimum of 1.25 eq. of anhydrous K_2CO_3 (0.27 g, 1.9 mmol). The starting material **36** (0.24 g, 0.83 mmol) was dissolved in a small amount of DMF (approx. 5 mL) and added to the reaction mixture. After stirring for 10 minutes, the reaction mixture was then brought to reflux temperature and allowed to reflux under an inert atmosphere for 16 hours. Upon completion of the reaction, water was added and the reaction mixture was extracted with four 30 mL portions of CH_2Cl_2 . The organic layers were combined and dried over anhydrous MgSO_4 . The solvent was removed under reduced pressure and the residue was adsorbed onto silica gel and purified via flash column chromatography. Analytical thin layer chromatography (TLC) indicated the desired product had an R_f value of approx. 0.75 when developed in 30:70 EtOAc/hexanes. Purification via flash column chromatography (on silica gel) was performed with 500ml of hexanes until any remaining starting material had completely eluted; then the polarity of the eluent mixture was increased to 10:90 EtOAc/hexanes to afford the desired product **10a** (0.24g, 0.57 mmol, 69%) as an opalescent oil. $^1\text{H-NMR}$ (CDCl_3): δ 7.31 (5H, m), 7.15 (1H, dd, $J = 8.1, 8.0$ Hz), 6.82 (7H, m), 5.27 (1H, dd, $J = 8.4, 3.2$ Hz), 3.19 (4H, dd, $J = 5.1, 5.0$ Hz), 3.04 (1H,

dd, $J = 13.8, 8.5$ Hz), 2.82 (2H, ddd, $J = 9.4, 8.1, 5.0$ Hz), 2.73 (3H, ddd, $J = 6.9, 6.9, 3.1$ Hz). ^{13}C -NMR (CDCl_3): δ 154.3, 152.7, 147 (faint), 140.6, 135.3, 130.4, 129.1, 129.1, 128.3, 126.5, 126.5, 119.7, 117.6, 117.5, 116.3, 116.1, 116.0, 114.3, 80.1, 65.7, 53.8, 53.8, 49.2, 49.2. ^{19}F -NMR (CDCl_3): δ -124.0. HRMS (EI) M^+ calculated for $\text{C}_{24}\text{H}_{24}\text{ClFN}_2\text{O}$: 410.91, found 410.16. Elemental analysis (%) calculated for $\text{C}_{24}\text{H}_{24}\text{ClFN}_2\text{O}$: C, 70.15; H, 5.89; N, 6.82; F, 4.62; Cl, 8.63. Found: C, 70.43; H, 6.02; N, 6.75; F, n/a; Cl, n/a.

6. Preparation of 1-[2-(4-fluorophenoxy)-2-phenylethyl]-4-(2-methoxyphenyl)-piperazine (11a)

[0053] *a. Formation of the free base form of 1-(2-methoxyphenyl)-piperazine.* The free base form of 1-(2-methoxyphenyl)-piperazine was formed by dissolving a liberal amount of the corresponding hydrochloride salt (**III**) in deionized water and adding an excess of 2 M aqueous NaOH. After 10 minutes, the aqueous mixture was extracted with four 30 mL portions of ethyl ether. The combined organic layers were dried over anhydrous MgSO_4 and the solvent was removed under reduced pressure. The secondary amine (free base form) of 1-(2-methoxyphenyl)-piperazine was isolated as an off-white solid and confirmed via NMR spectroscopy. ^1H -NMR (CDCl_3): δ 7.07 (1H, ddd, $J = 9.1, 8.0, 4.5$ Hz), 6.94 (1H, d, $J = 8.6$ Hz), 6.93 (1H, d, $J = 4.1$ Hz), 6.89 (1H, d, $J = 7.7$ Hz), 3.87 (3H, s), 3.38 (4H, d, $J = 15.3$ Hz), 3.38 (2H, d, $J = 4.0$ Hz), 3.14 (1H, bs), 2.85 (1H, bs), 1.83 (1H, s). ^{13}C -NMR (CDCl_3): δ 152.2, 141.7, 122.7, 120.9, 118.1, 111.1, 55.2, 52.0, 46.3, 46.3 Melting point: 180-183 °C (dec.)

[0054] *b. Formation the coupled product 11a.* To a well-stirred solution of approx. 1.6 eq. of 1-(2-methoxyphenyl)-piperazine (**III**) (0.34 g, 1.78 mmol) in 20 mL DMF at 25°C under N_2 and fitted with a reflux condenser was added a minimum of 1.25 eq. of anhydrous K_2CO_3 (0.34 g, 2.45 mmol). The starting material **36** (0.245 g, 0.83 mmol) was dissolved in a small amount of DMF (approx. 5 mL) and added to the reaction mixture. After stirring for 10 minutes, the reaction mixture was then brought to reflux temperature and allowed to reflux under an inert atmosphere for 16 hours. Upon completion of the reaction, water was added and the reaction mixture was extracted with four 30 mL portions of CH_2Cl_2 . The organic layers were combined and dried over anhydrous MgSO_4 . The solvent was removed under reduced pressure and the residue was adsorbed onto silica gel and purified via flash column chromatography. Analytical thin layer chromatography (TLC) indicated the desired product had an R_f value of approx. 0.7 when developed in 30:70 EtOAc/hexanes. The flash column chromatography (on

silica gel) was performed with 500mL of hexanes until any remaining starting material had completely eluted; then the polarity of the eluent mixture was increased to 10:90 EtOAc/hexanes to afford the desired product **11a** (0.114 g, 0.28 mmol, 34%) as an opalescent oil. $^1\text{H-NMR}$ (CDCl_3): δ 7.32 (5H, m), 6.9 (8H, m), 5.30 (1H, dd, $J = 8.5, 3.1$ Hz), 3.86 (3H, s), 3.1 (5H, dd, $J = 13.7, 8.5$ Hz), 2.8 (5H, m). $^{13}\text{C-NMR}$ (CDCl_3): δ 152.6, 152.6, 141.7, 140.9, 129.1, 129.1, 129.1, 128.2, 126.5, 126.5, 123.3, 121.4, 118.6, 117.7, 117.6, 116.2, 116.0, 111.5, 79.9, 66.0, 55.7, 55.7, 54.2, 51.1, 51.1. $^{19}\text{F-NMR}$ (CDCl_3): δ -124.2. HRMS (EI) M^+ calculated for $\text{C}_{25}\text{H}_{27}\text{FN}_2\text{O}_2$: 406.49; found 406.21. Elemental analysis (%) calculated for $\text{C}_{25}\text{H}_{27}\text{FN}_2\text{O}_2$: C, 73.87; H, 6.69; F, 4.67; N, 6.89. Found: C, 73.84; H, 6.95; F, n/a; N, 6.92.

7. Preparation of 1-[2-(4-fluorophenoxy)-2-phenylethyl]-4-(3-trifluoromethylphenyl)-piperazine (12a)

[0055] To a well-stirred solution of approx. 1.1 eq. of m-trifluoromethylphenyl-piperazine (**IV**) (0.44 g, 1.9 mmol) in 20mL DMF at 25 °C under N_2 and fitted with a reflux condenser was added a minimum of 1.25 eq. of anhydrous K_2CO_3 (0.33 g, 2.4 mmol). The starting material **36** (0.41 g, 1.4 mmol) was dissolved in a small amount of DMF (approx. 5 mL) and added to the reaction mixture. After stirring for 10 minutes, the reaction mixture was then brought to reflux temperature and allowed to reflux under an inert atmosphere for 16 hours. Upon completion of the reaction, water was added and the reaction mixture was extracted with four 30 mL portions of CH_2Cl_2 . The organic layers were combined and dried over anhydrous MgSO_4 . The solvent was removed under reduced pressure and the residue was adsorbed onto silica gel and purified via flash column chromatography. Analytical thin layer chromatography (TLC) indicated the desired product had an R_f value of approx. 0.75-0.8 when developed in 30:70 EtOAc/hexanes. Purification via flash column chromatography (on silica gel) was performed with 500 mL of hexanes until any remaining starting material had completely eluted; then the polarity of the eluent mixture was increased to 10:90 EtOAc/hexanes to afford the desired product **12a** (0.11 g, 0.25 mmol, 18%) as an opalescent oil. $^1\text{H-NMR}$ (CDCl_3): δ 7.32 (6H, m), 7.09 (1H, d, $J = 7.6$ Hz), 7.07 (1H, d, $J = 7.8$ Hz), 7.05 (1H, d, $J = 7.0$ Hz), 6.88 (1H, d, $J = 9.3$ Hz), 6.85 (1H, d, $J = 8.2$ Hz), 6.80 (1H, d, $J = 9.3$ Hz), 6.79 (1H, d, $J = 9.3$ Hz), 5.28 (1H, dd, $J = 8.4, 3.3$ Hz), 3.24 (4H, dd, $J = 5.0, 5.0$ Hz), 3.05 (1H, dd, $J = 13.9, 8.5$ Hz), 2.84 (2H, ddd, $J = 10.8, 5.2, 5.2$ Hz), 2.74 (3H, m). $^{13}\text{C-NMR}$ (CDCl_3): δ 152 (faint), 151 (faint), 145 (faint), 140 (faint), 129.5, 129.5, 128.7, 128.7, 127.9, 126.1, 126.1, 118.6, 117.2, 117.1, 115.8, 115.8, 115.6, 112, 112, 79.7, 65.2, 53.4, 53.4, 48.7, 48.7. $^{19}\text{F-NMR}$ (CDCl_3): δ -63.1, -

123.5. HRMS (EI) M^+ calculated for $C_{25}H_{24}F_4N_2O$: 444.46, found 444.18. Elemental analysis (%) calculated for $C_{25}H_{24}F_4N_2O$: C, 67.56; H, 5.44; F, 17.10; N, 6.30. Found: C, 68.20; H, 5.82; F, 15.22; N, 5.90.

8. Formation of the hydrochloride salt 10b from piperazinyl-containing base 10a

[0056] To a well-stirred solution of **10a** (0.235 g, 0.57 mmol) in 30mL of anhydrous THF was added 10 eq. of 2M HCl in anhydrous diethyl ether (3 mL, 6.0 mmol) at 25 °C under N_2 . The mixture was allowed to stir for approx. 12 to 16 hours. Upon completion of the reaction, the reaction mixture was filtered and the filtered residue was washed with freshly distilled CH_2Cl_2 . The solute was then allowed to dry and was subsequently re-dissolved in approx. 5 mL of anhydrous EtOH. The resulting solution was refluxed briefly for 5-10 minutes and the solvent then was removed under reduced pressure. The remaining solid was placed under vacuum until all residual solvent was removed affording the desired hydrochloride salt **10b** as a greenish crystalline solid (0.17 g, 0.39mmol, 68%). Confirmation of the HCl salt **10b** was made by treating a small amount of the final product with strong base, extracting with ethyl ether, removing the organic solvent under reduced pressure, and confirming via NMR using $CDCl_3$. NMR data corresponded with those results previously reported for the free base form **10a**. TLC was also performed (30:70 EtOAc/hexanes) and the R_f value of approx. 0.75 matched that of **10a**. Reverse phase TLC was performed on the HCl salt **10b** and R_f values of 0.3 in 100% CH_3CN and 0.8 in 2:98 H_2O/CH_3CN were recorded, respectively.

9. Formation of the hydrochloride salt 11b from piperazinyl-containing base 11a

[0057] To a well-stirred solution of **11a** (0.114g, 0.28 mmol) in 30mL of anhydrous THF was added 10 eq. of 2 M HCl in anhydrous diethyl ether (1.6 mL, 3.2 mmol) at 25 °C under N_2 . The mixture was allowed to stir for approx. 12 to 16 hours. Upon completion of the reaction, the reaction mixture was filtered and the filtered residue was washed with freshly distilled CH_2Cl_2 . The solute was then allowed to dry and was subsequently re-dissolved in approx. 5 mL of anhydrous EtOH. The resulting solution was refluxed briefly for 5-10 minutes and the solvent was then removed under reduced pressure. The remaining solid was placed under vacuum until all residual solvent was removed affording the desired hydrochloride salt **11b** as a white crystalline solid (0.0112 g, 0.025 mmol, 9%), albeit in low yield. Confirmation of the HCl salt **11b** was made by treating a small amount of the final product with strong base, extracting with ethyl ether, removing the organic solvent under reduced pressure, and confirming

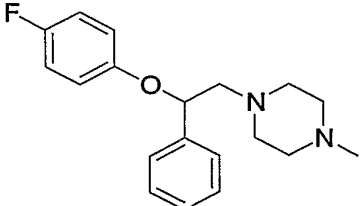
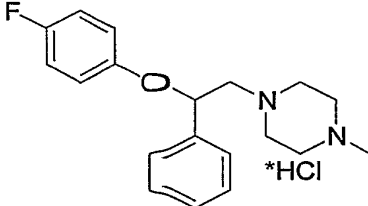
via NMR using CDCl_3 . NMR data corresponded with those results previously reported for the free base form **11a**. TLC was also performed (30:70 EtOAc/hexanes) and the R_f value of approx. 0.7 matched that of **11a**. Reverse phase TLC was performed on the HCl salt **11b** and R_f values of 0.25 in 100% CH_3CN and 0.75 in 2:98 $\text{H}_2\text{O}/\text{CH}_3\text{CN}$ were recorded, respectively.

10. Formation of the hydrochloride salt **12b** from piperazinyl-containing base **12a**

[0058] To a well-stirred solution of **12a** (0.11 g, 0.25 mmol) in 30 mL of anhydrous THF was added 10 eq. of 2 M HCl in anhydrous diethyl ether (1.4 mL, 2.8 mmol) at 25 °C under N_2 . The mixture was allowed to stir for approx. 12 to 16 hours. Upon completion of the reaction, the reaction mixture was filtered and the filtered residue was washed with freshly distilled CH_2Cl_2 . The solute was then allowed to dry and was subsequently re-dissolved in approx. 5 mL of anhydrous EtOH. The resulting solution was refluxed briefly for 5-10 minutes and the solvent then was removed under reduced pressure. The remaining solid was placed under vacuum until all residual solvent was removed affording the desired hydrochloride salt **12b** (0.0496 g, 0.103 mmol, 42%). Confirmation of the HCl salt **12b** was made by treating a small amount of the final product with strong base, extracting with ethyl ether, removing the organic solvent under reduced pressure, and confirming via NMR using CDCl_3 . NMR data corresponded with those previously reported for the free base form **12a**. TLC was also performed (30:70 EtOAc/hexanes) and the R_f value of approx. 0.75-0.8 matched that of **12a**. Reverse phase TLC was performed on the HCl salt **12b** and R_f values of 0.35 in 100% CH_3CN and 0.85 in 2:98 $\text{H}_2\text{O}/\text{CH}_3\text{CN}$ were recorded, respectively.

EXAMPLE 5. SYNTHESIS OF COMPOUNDS 13 – 16

[0059] The following compounds were synthesized according to Scheme 4 shown in Figure 8 and the procedures described below:

Cpd	Structure (Free Amine)	Cpd	Structure (Hydrochloride Salt)
13a		13b	

14a		14b	
15a		15b	
16a		16b	

1. General Procedure for preparation of 2-amino-1-phenylethanol (38 and 39): Synthesis of 2-(4-Methyl-piperazin-1-yl)-1-phenylethanol (38)

[0060] To a solution of 3.0 equivalents of 1-methylpiperazine (VI) (25.0g, 24.96 mmol) in 100 mL was added 3.0 equivalents of K₂CO₃ (34.5g, 24.96 mmol) and the solution allowed to stir for 10 minutes. After that time, styrene oxide 37 (10.0g, mmol) was added to the solution and the reaction mixture was refluxed overnight. After 16h, the reaction mixture was cooled to room temperature and quenched with 200mL of water. The reaction was extracted with dichloromethane (4x100mL) and the organic layer washed with brine. The organic layer was dried over Na₂SO₄ and the solvent evaporated. The product was purified using flash chromatography, eluting pure ethyl acetate until any remaining starting material had been recovered and then increasing the polarity of the solvent using 30:70 methanol/ethyl acetate. The product 38 (5.66g, 25.73 mmol, 31%) was afforded as a white-pink solid. ¹H-NMR (CDCl₃): δ 7.36 (5H, m), 4.74 (1H, dd, *J* = 4.05, 4.06 Hz), 4.00 (1H, b), 2.79 (2H, b), 2.51 (8H, m), 2.31 (3H, s).

2. General Procedure for Preparation of 1-[2-(4-aryloxy)-2-phenyl-ethyl]-4-aryl-piperazines (13a to 16a)

a. Preparation of 1-[2-(4-Fluoro-phenoxy)-2-phenyl-ethyl]-4-methyl-piperazine (13a)

[0061] 2-(4-Methyl-piperazin-1-yl)-1-phenyl-ethanol (**38**) (2.0g, 9.09 mmol) was dissolved in 50mL of THF. 4-fluorophenol was added to the solution and then triphenylphosphine. After complete dissolution, the mixture was cooled to 0°C and after 10 minutes DIAD was added dropwise. Ice-bath was removed and the reaction mixture was stirred at room temperature for 48 hours. The product was purified using flash chromatography, eluting pure ethyl acetate to recover any starting materials. Polarity of the solvent was then increased using 30:70 methanol/ethyl acetate. The product **13a** (2.18g, 6.94 mmol, 76%) was afforded as a orange-brown oil. ¹H-NMR (CDCl₃): δ 7.33 (5H, m), 6.80 (4H, m), 5.21 (1H, dd, *J* = 3.08, 3.07 Hz), 3.00 (1H, dd, *J* = 8.49, 8.48), 2.67 (5H, b), 2.45 (4H, b), 2.27 (3H, s).

b. Preparation of 1-Methyl-4-[2-phenyl-2-(4-trifluoromethyl-phenoxy)-ethyl]-piperazine (14a)

[0062] 2-(4-Methyl-piperazin-1-yl)-1-phenyl-ethanol (**38**) (2.0g, 9.09 mmol) was dissolved in 50mL of THF. 4-trifluoromethyl phenol was added to the solution and then triphenylphosphine. After complete dissolution, the mixture was cooled to 0°C and after 10 minutes DIAD was added dropwise. Ice-bath was removed and the reaction mixture was stirred at room temperature for 48 hours. The product was purified using flash chromatography, using pure ethyl acetate. The product **14a** (2.78g, 7.64 mmol, 84%) was afforded as a yellow oil. H-NMR (CDCl₃): δ 7.39 (2H, d, *J* = 8.62), 7.36 (5H, m), 6.91 (2H, d, *J* = 8.58), 5.38 (1H, dd, *J* = 3.11, 3.10 Hz), 3.03 (1H, dd, *J* = 8.39, 8.38), 2.71 (1H, dd, *J* = 3.14, 3.14), 2.63 (4H, b), 2.44 (4H, b), 2.27 (3H, s).

3. General Procedure for Preparation of the Hydrochloride Salts 13b to 16b from free amines 13a to 16a: Preparation of Hydrochloride salt 13b from free amine 13a

[0063] To a solution of 1.0 g of product **13a** (3.18 mmol) in 10 mL of stirring THF under N₂, 10 equivalents of 2M HCl in anhydrous diethyl ether (15.9mL, 31.8 mmol) was added at room temperature. The mixture was allowed to stir for 12 to 16 hours. The reaction mixture was then filtered and redissolved in 10 mL of anhydrous ethyl alcohol. The reaction was heated to reflux conditions for 5 to 10 minutes, and the solvent was then evaporated by reduced pressure

yielding 0.83 g of the final product **13b** (2.38 mmol, 75%) as a white-cream solid. The hydrochloride salt product was confirmed by treating a sample of the product with strong base, extracting with diethyl ether, evaporating the organic layer under reduced pressure, and confirming by NMR in CDCl₃. The NMR data corresponded with the free amine product **13a**.

EXAMPLE 6. [³H] 5-HT UPTAKE ASSAY

[0064] A [³H]5-HT uptake assay was performed to determine the ability of the 2-(4-fluorophenoxy)-2-phenyl-ethyl piperazines (Compounds 10b, 11b, and 12b) to inhibit [³H]5-HT uptake (Dorsey et al., 2004). The specific uptake of [³H]5-HT by human platelets was typically > 90% specific as defined by 10 μM fluoxetine.

[0065] Outdated human platelets were obtained from the blood bank at Pitt County Memorial Hospital, Greenville, NC. Platelets from 5-10 donors were pooled, 10% dimethylsulfoxide was added, and aliquots were stored frozen at -80°C until use. For assays, 5 ml of platelets were thawed and suspended in 20 mL ice-cold Krebs-Ringer-HEPES (KRH) buffer as previously described (Cozzi and Foley, 2002). The KRH buffer contained the following ingredients: 124.0 mM NaCl, 2.9 mM KCl, 1.3 mM MgSO₄, 1.2 mM KH₂PO₄, 2.4 mM CaCl₂, 5.2 mM *d*-glucose, 25.0 mM HEPES, 0.1 mM sodium ascorbate, 0.1 mM pargyline. The buffer was adjusted to pH 7.4 with 5 M NaOH. The ability of platelets to accumulate [³H]serotonin was measured in the absence and presence of test drugs as follows: a 490 μl aliquot of the platelet suspension was added to glass tubes containing 5 μl test drug (various concentrations, dissolved in DMSO), 5 μl DMSO (for total determinations), or 5 μl fluoxetine hydrochloride dissolved in DMSO (for nonspecific determinations; final concentration, 10 μM). The assay tubes were preincubated in a 37 °C shaking water bath for 5 min. The tubes were then returned to the ice bath and chilled for 10 min. [³H]Serotonin was added (5 μl of stock solution; final concentration, 10 nM), giving a total incubation volume of 500 μl. All tubes were returned to the 37 °C shaking water bath for 5 min to initiate neurotransmitter uptake. Uptake was terminated by chilling the test tubes in the ice bath. After adding 3 ml ice-cold KRH, each assay tube was immediately vacuum filtered through glass fiber filters (Whatman GF/B) pretreated with 0.1% polyethyleneimine. Filters were washed with 2 x 3 ml ice-cold KRH, allowed to dry briefly under vacuum, then placed in liquid scintillation vials. Scintillation cocktail (8 ml) was added and the vials were sealed, vortexed, and allowed to stand overnight. Radioactivity was measured using liquid scintillation spectroscopy (Packard Tri-Carb 1600 CA). Specific uptake

was defined as uptake at 37 °C minus uptake in the presence of 10 μ M fluoxetine. Under these conditions, specific [3 H]serotonin uptake was typically 90 % of total uptake. The IC₅₀ value for each test drug was determined from displacement curves using at least 6 drug concentrations, each run in triplicate. Data were transformed from dpm to % specific uptake and fitted to a four-parameter logistic curve using commercial computer software, from which the IC₅₀ values are obtained. The IC₅₀ values are shown in Table 1 below.

Table 1

CMPD % #	IC ₅₀ (μ M) SERT
10b	1.45
11b	3.27
12b	9.56

[0066] Dose-response curves for inhibition of [3 H]5-HT uptake by Compounds **10b**, **11b**, and **12b** are presented in Figures 9 – 11. The inhibition curves all had slope coefficients that did not differ from unity, indicating that the compounds all interacted with a single site on the serotonin reuptake transporter (SERT). Compounds **10b** and **11b** were the most potent inhibitors of [3 H]5-HT accumulation, with IC₅₀ values of $1.45 \pm 0.08 \mu$ M and $3.27 \pm 0.23 \mu$ M, respectively. These values did not differ from each other at the $P < 0.05$ level of significance (one-way ANOVA followed by a Student-Newman-Keuls t -test). Compound **12b** was the least potent compound in the series with an IC₅₀ value of $9.56 \pm 2.4 \mu$ M; this was only one-sixth the potency of compound **10b** and three-fold less potent than compound **11b** ($P < 0.05$). It is apparent that the new compounds are significantly less potent than fluoxetine, which exhibits nanomolar (nM) affinity for the SERT, but it is not possible at this stage to decide whether the decreased potency is due to electrostatic, stereochemical, or steric considerations or some combination of these. For example, the phenylpiperazine moiety of the new compounds adds considerable bulk to the side-chain amine of fluoxetine and also changes its electronic character with the addition of new lone pair electrons and an electron-rich aromatic ring. Without wanting to be bound by theory, it is possible that the electron-withdrawing, hydrophobic trifluoromethyl group present on the phenyl ring of **12b** is responsible for the attenuated serotonin uptake inhibition activity of this compound when compared to compounds **10b** and **11b**, but the methoxy substitution on compound **11b** also represents a positional isomer and this may influence binding through a steric mechanism.

EXAMPLE 7. BIOLOGICAL EVALUATION

[0067] Biological Evaluation of selected compounds was done through the National Institutes of Health Psychoactive Drug Screening Program (NIH-PDSP). Biological evaluation of the synthesized molecules consisted of: (a) Neurotransmitter uptake assays for the human serotonin transporter (SERT), (b) Binding assays for the serotonin receptor 5-HT_{1A}, (c) Binding assays for the serotonin receptor 5-HT_{2A}, and (d) Comparison of the biological data obtained for the synthesized molecules with that of currently used psychotherapeutics.

[0068] All binding assays were carried out with an incubation time of 60 minutes at room temperature. All binding assays were performed in a similar manner, although the buffers used differ from assay to assay.

a. Binding assays for the 5-HT Receptor Subtypes

[0069] Table 2 below gives the (1) radioligand; (2) assay buffer; (3) unlabelled reference ligand; and (4) references to the commonly-known procedures which were used.

Table 2

Serotonin	Radioligand	Assay buffer	Unlabelled ligand as reference compound	References
5-HT1A	3H-8-OH-DPAT	A	5-HT	Roth lab standard protocol
5-HT2A	3H-ketanserin	A	Chlorpromazine	Roth lab Standard protocol

[0070] The assay buffer A was made up of 50 mM Tris-Cl, 0.1 mM EDTA, and 10 mM MgCl₂, at a pH of 7.4. To perform the assay, a 96-well plate was set up as follows:

UNK #1	UNK #1	UNK #2	UNK #2	UNK #3	UNK #3	UNK #4	UNK #4
UNK #5	UNK #5	UNK #6	UNK #6	UNK #7	UNK #7	UNK #8	UNK #8
UNK #9	UNK #9	UNK #10	UNK #10	UNK #11	UNK #11	UNK #12	UNK #12
STD #1	STD #1	STD #2	STD #2	TOTAL BINDING	TOTAL BINDING	TOTAL BINDING	TOTAL BINDING
STD #1 AT 1 nM	STD #1 AT 1 nM	STD #1 AT 3 nM	STD #1 AT 3 nM	STD #1 AT 10 nM	STD #1 AT 10 nM	STD #1 AT 30 nM	STD #1 AT 30 nM
STD #1 AT 100 nM	STD #1 AT 100 nM	STD #1 AT 300 nM	STD #1 AT 300 nM	STD #1 AT 1000 nM	STD #1 AT 1000 nM	TOTAL COUNTS ADDED / PLATE	TOTAL COUNTS ADDED / PLATE

[0071] For the 3H-ligands 1-2 nM final concentration of radioligand was used. For 125I-radioligands, 0.05-0.1 nM final concentration was used. The following components were pipetted in the following order: binding buffer, radioligand, cold unknown ligand, cold reference ligand, and membranes. The wells were incubated at room temperature for 60 minutes. At the end of the incubation, the samples were harvested onto pre-soaked (0.3% polyethyleneimine) GF/C filters using a 96-well harvester. They were then washed with 3 quick washes using ice-cold harvest buffer. The filters were removed and air dried overnight. Samples were counted in a 96-well counter using 100 µl of scintillant/well. The K_i value of the reference compound was calculated using the LIGAND Program (Munson and Rodbard, 1980). The percent inhibition was calculated using the following formula:

$$\text{Inhibition} = \frac{\text{Counts bound at 10 uM concentration of unknown compound}}{\text{Total specific counts}} \times 100$$

b. Binding assays for the Transporters

[0072] Representative K_i values for transporters have already been determined as follows:

5-HTT: Imipramine $K_i = 12 \pm 2$ nM
 VMAT: Ketanserin $= 20 \pm 1$ nM
 NET: Nisoxetine $= 1 \pm 0.5$ nM

[0073] Table 3 below gives the (1) radioligand; (2) assay buffer; and (3) unlabelled reference ligand.

Table 3

Transporter	Radioligand	Assay buffer	Unlabelled ligand as reference compound
SERT	3H-Citalopram	H	Fluoxetine

[0074] The assay buffer H contained 50 mM Tris-HCl, 150 mM NaCl, and 5mM KCl, at a pH of 7.4. For the peptide receptor binding assay, filters were pre-soaked for 7 days in 0.3% PEI. To perform the assay, a 96-well plate was set up as follows:

UNK #1	UNK #1	UNK #2	UNK #2	UNK #3	UNK #3	UNK #4	UNK #4
UNK #5	UNK #5	UNK #6	UNK #6	UNK #7	UNK #7	UNK #8	UNK #8
UNK #9	UNK #9	UNK #10	UNK #10	UNK #11	UNK #11	UNK #12	UNK #12
STD #1	STD #1	STD #2	STD #2	TOTAL BINDING	TOTAL BINDING	TOTAL BINDING	TOTAL BINDING
STD #1 AT 1 nM	STD #1 AT 1 nM	STD #1 AT 3 nM	STD #1 AT 3 nM	STD #1 AT 10 nM	STD #1 AT 10 nM	STD #1 AT 30 nM	STD #1 AT 30 nM
STD #1 AT 100 nM	STD #1 AT 100 nM	STD #1 AT 300 nM	STD #1 AT 300 nM	STD #1 AT 1000 nM	STD #1 AT 1000 nM	TOTAL COUNTS ADDED / PLATE	TOTAL COUNTS ADDED / PLATE

[0075] For the 3H-ligands 1-2 nM final concentration of radioligand was used. For 125I-radioligands, 0.05-0.1 nM final concentration was used. The following components were

pipetted in the following order: binding buffer, radioligand, cold unknown ligand, cold reference ligand, and membranes. The wells were incubated at room temperature for 60 minutes. At the end of the incubation, the samples were harvested onto pre-soaked (0.3% polyethyleneimine) GF/C filters using a 96-well harvester. They were then washed with 3 quick washes using ice-cold harvest buffer. The filters were removed and air dried overnight. Samples were counted in a 96-well counter using 100 μ l of scintillant/well. The K_i value of the reference compound was calculated using the LIGAND Program (Munson and Rodbard, 1980). The percent inhibition was calculated using the following formula:

$$\text{Inhibition} = \frac{\text{Counts bound at 10 uM concentration of unknown compound}}{\text{Total specific counts}} \times 100$$

[0076] In Table 4 below, data represent mean % inhibition (N = 4 determinations) for compounds tested at receptor subtypes. In Table 53 below, Data represent K_i Values. K_i values are reported in nanomolar concentration. Significant inhibition is considered >50%. If the result was <50%, NIH-PDSP did not provide the information.

Table 4

CMPD #	5-HT1A	5-HT2A	SERT
1b	100.0	96.0	62.0
2b	91.0	81.0	67.0
3b	98.0	61.0	12.0
4b	93.0	7.0	16.0
5b	71.0	17.0	22.0
6b	102.0	11.0	-1.0
7b	76.0	-2.0	16.0

Table 5

CMPD #	Ki (nM)		
	5-HT _{1A}	5-HT _{2A}	SERT
1b	102.0	237.1	1,242.00
2b	716.1	195.2	1,871.00
3b	257.9	3,493.0	-
4b	375.3	-	-
5b	712.5	-	-
6b	88.7	-	-
7b	724.4	-	-
8-OH-DPAT	1.2 ^a	> 1000 ^a	> 1000 ^a
mirtazapine	> 1000 ^a	14.8 ^a	> 1000 ^a
fluoxetine	> 1000 ^a	> 1000 ^a	30.8 ^a

^a Orjales, et al. 2003.

[0077] Ten (10) of the sixteen (16) compounds synthesized were biologically evaluated. Compounds 10b, 11b and 12b were previously evaluated for their serotonin uptake ability as described in Example 5. For screening purposes, compounds 1b through 7b were analyzed for their affinity towards the serotonin 5-HT_{1A} receptor as well as their affinity towards the 5-HT_{2A} receptor and the SERT. Data in Table 2 represent mean % inhibition (N = 4 determinations) for compound tested at receptor subtypes. Significant inhibition is considered > 50%. In cases where negative inhibition (-) is seen, this represents a stimulation of binding. Occasionally, compounds at high concentrations will non-specifically increase binding. Ki values were determined only for compounds with % inhibition > 50%. As can be seen in Table 3, all the compounds show low to moderate affinity for 5-HT_{1A}. Compound **6b** shows the highest affinity for 5-HT_{1A}. Compounds **1b** and **2b** exhibit the desired dual activity towards the 5-HT_{2A} receptor and the SERT, although low. None of the compounds has exhibited significant affinity for SERT; however, compounds **1b** and **2b** show low affinity which is still higher than that showed by compounds **10b** through **12b** (see Table 1 in Example 5).

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The entire content of each of the references given below is hereby incorporated by reference.

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